

ANHYDROBARTOGENIC ACID AND 19-EPIBARTOGENIC ACID, TWO TRITERPENES FROM *BARRINGTONIA SPECIOSA*

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Key Word Index—*Barringtonia speciosa*; Barringtoniaceae; triterpenoid acids; anhydrobartogenic acid; 19-epibartogenic acid.

Abstract—Two new triterpene dicarboxylic acids, isolated from the fruits of *Barringtonia speciosa*, have been identified as anhydrobartogenic acid and 19-epibartogenic acid by spectral and chemical evidence.

INTRODUCTION

Recently, we reported [1] the isolation of a new triterpene, bartogenic acid, from *Barringtonia speciosa* Forst and established its structure as $2\alpha,3\beta,19\alpha$ -trihydroxyolean-12-en-24,28-dioic acid (**1**). In continuation of our studies on the minor constituents from the fruits of *B. speciosa*, we have isolated two new triterpene dicarboxylic acids, isolated as their dimethyl esters, BS-1 and BS-3, and the present paper describes their structures.

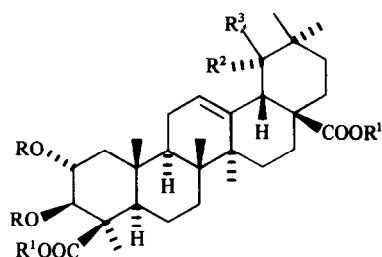
RESULTS AND DISCUSSION

Compound BS-1, mp 270–272°, $C_{32}H_{48}O_6$ ($[M]^+$ at m/z 528); UV λ_{\max} 242 nm (ϵ 7900); IR ν_{\max} 3500 (hydroxyl) and 1725 (ester) cm^{-1} ; 1H NMR ($CDCl_3$): δ 0.78, 0.85, 0.86, 0.95, 0.97, 1.25 (s, 3H each, $6 \times Me$), 3.61, 3.69 (s, $2 \times COOMe$), 2.92 (*br d*, $J = 10$ Hz), 4.07 (*m*, 1H), 5.40 (s, 1H) and 5.48 (*m*, 1H). From the UV data, BS-1 was considered to be a heteroannular diene. Acetylation with acetic anhydride–pyridine afforded a diacetate (**9**), $C_{36}H_{52}O_8$, mp 140–142°, $[\alpha]_D + 80^\circ$ ($CHCl_3$), λ_{\max} 242 nm (ϵ 7900). The 1H NMR spectrum indicated the presence of two acetoxy (δ 1.98 and 2.08) and the protons geminal to the acetoxy appeared at δ 4.84 (*d*, $J = 10$ Hz) and 5.70 (*m*) in *trans*-diaxial orientation as in diacetyl dimethyl bartogenate (**2**) [1, 2]. Thus the two acetoxy in BS-1 diacetate (**9**) are *trans*-diequatorial. The low-field signal for 2β -H (δ 5.70) indicated that it is in the carbonyl zone of a carboxymethyl group. The carboxymethyl group is β -axial at C-4 as in diacetyl dimethyl bartogenate (**2**). The presence of a 12,18-diene system is inferred from its 1H NMR spectral data since the vinyl protons appeared at δ 5.40 and 5.48. If the 11,13(18)-diene system were present in the molecule, the signals for the olefinic protons would have appeared as an ABX pattern, the 12-H being deshielded very much downfield (δ 6.5). Furthermore, the signal for 19-H appeared as a singlet at δ 5.40. Isomerization of BS-1 diacetate with methanol–hydrochloric acid afforded the stable transoid diene (**10**) [1, 2], having the typical UV absorption spectrum [λ_{\max} 244, 252, 261 nm (ϵ 19 559, 25 120, 16 600)] characteristic of the 11,13(18)-diene system [3], which was identified by comparison with an authentic specimen obtained from diacetyl dimethyl bartogenate. This con-

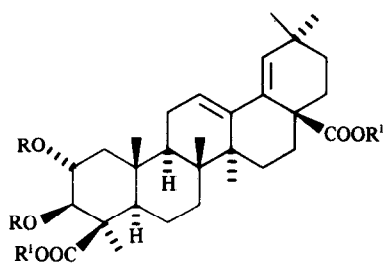
clusively proves that BS-1 has a 12,18-diene system and is therefore assigned the structure dimethyl- $2\alpha,3\beta$ -dihydroxyolean-12,18-diene-24,28-oate (**8**). Corroborative evidence for the presence of a Δ^{18} -double bond was obtained from the mass spectrum of BS-1 which showed an ion at m/z 260 for the fragment **11** due to retro-Diels–Alder fission [4].

Compound BS-3, $C_{32}H_{50}O_7$, mp 252–254°, $[\alpha]_D + 100^\circ$ ($CHCl_3$); UV λ_{\max} 205 nm; IR ν_{\max} 3500, 1725, 1700 cm^{-1} ; 1H NMR ($CDCl_3$): δ 0.77, 0.82, 0.90, 1.06, 1.16, 1.25 (s, 3H each, $6 \times Me$), 2.59 (*d*, 1H, $J = 11$ Hz), 3.36 (*d*, $J = 11$ Hz), 3.49 (*d*, 1H, $J = 10$ Hz), 3.63 (s, 3H, $COOMe$), 3.68 (s, 3H, $COOMe$), 4.0 (*m*, 1H), 5.45 (*m*, 1H) is assigned structure **4** on the basis of the following evidence.

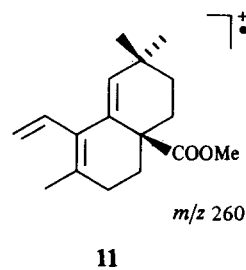
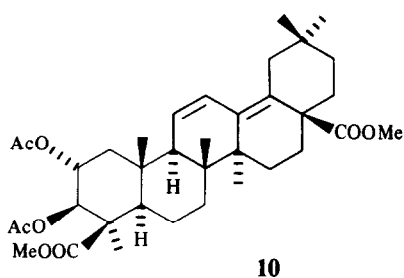
BS-3 formed a diacetate (**5**), $C_{36}H_{54}O_9$, ($[M]^+$ at m/z 630), mp 138–140°, whose IR spectrum showed the presence of a free hydroxyl (3540 cm^{-1}). On treatment with perchloric acid–acetic anhydride, it formed an amorphous triacetate. The 1H NMR spectrum of BS-3 diacetate showed the presence of hydrogens geminal to the acetoxy in a *trans*-diaxial system (δ 4.85, *d*, $J = 10$ Hz; 5.69, *m*). The presence of a carboxymethyl group at C-4 in the axial position caused a downfield shift of the 2β -H, as in the case of BS-1 diacetate. A high-field signal observed for one of the methyl groups indicated the presence of a methoxy carbonyl group at C-17. The presence of a methoxy carbonyl in ring A at C-4 caused the C-23 methyl to resonate at low field (δ 1.25). The positions of the acetoxy and carboxymethyls in BS-3 diacetate having been established, the position of the hindered hydroxyl group remained to be settled. The mass spectrum of BS-3 and its diacetate showed an intense peak at m/z 278 due to retro-Diels–Alder fragmentation [4], suggesting that the hindered hydroxyl group is present in ring D or E. Oxidation of BS-3 diacetate with CrO_3 –pyridine produced a ketone (**6**), mp 236–238°, which was found to be identical to the ketone obtained [1, 2] from oxidation of diacetyl dimethyl bartogenate (**2**). Thus in BS-3 diacetate the hydroxyl group is present at C-19, and has β -configuration. This is supported by the 1H NMR spectral data; the 19α -H appeared as a broad doublet at δ 3.36 ($J = 11$ Hz) and the 18β -H as a doublet at 2.59 ($J = 11$ Hz). Thus 18β -H and 19α -H are *trans*-



- 1** $R = R^1 = R^3 = H$; $R^2 = OH$
2 $R = Ac$; $R^1 = Me$; $R^2 = OH$; $R^3 = OH$
3 $R = R^1 = R^2 = H$; $R^3 = OH$
4 $R = H$; $R^1 = Me$; $R^2 = H$; $R^3 = OH$
5 $R = Ac$; $R^1 = Me$; $R^2 = H$; $R^3 = OH$
6 $R = Ac$; $R^1 = Me$; $R^2 R^3 = O$



- 7** $R = R^1 = H$
8 $R = H$; $R^1 = Me$
9 $R = Ac$; $R^1 = Me$



diaxial to each other and therefore the hydroxyl has the β -configuration. Thus the structure of BS-3 is confirmed as dimethyl 2 α ,3 β ,19 β -trihydroxyolean-12-en-24,28-dioate (**4**).

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